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[Exergy - Wikipedia](#)

In thermodynamics, the exergy of a system is the maximum useful work possible during a process that brings the system into equilibrium with a heat reservoir. When the surroundings are the reservoir, exergy is the potential of a system to cause a change as it achieves equilibrium with its environment.

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Exergie bezeichnet den Teil der Gesamtenergie eines Systems, der Arbeit verrichten kann, wenn dieses in das thermodynamische Gleichgewicht mit seiner Umgebung gebracht wird. Exergie ist ein Potential zwischen mindestens zwei Zuständen, wobei einer davon meist der Umgebungszustand ist. Die Exergie ist im Gegensatz zur Energie keine

[What Is Exergy? Exergy Economics](#)

Exergy is a thermodynamic concept, used for many years within engineering analyses of chemical and mechanical processes and systems. Officially, exergy is defined as: The maximum useful work which can be extracted from a system as it reversibly comes into equilibrium with its environment. In other words, it is the capacity of energy to do

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The theory of exergy aims to develop an integrated analysis method that includes the first two laws of thermodynamics, and thus allows us to take into account both the amount of energy put into play and its quality, which the first law cannot do.

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What is the difference between exergy and energy? - Quora

Exergy is the energy that is available to be used. After the system and surroundings reach equilibrium, the exergy is zero. Determining exergy was also the first goal of thermodynamics.